The Dymola Thermo-Bond-Graph Library

- In this lecture, we shall introduce a third bond-graph library, one designed explicitly to deal with *convective flows*.
- To this end, we shall need to introduce a new type of bonds, bonds carrying in parallel three distinct, yet inseparable, power flows: a *heat flow*, a *volume flow*, and a *mass flow*.
- These new *bus-bonds*, together with their corresponding *bus-0-junctions*, enable the modeler to describe convective flows at a high abstraction level.
- The example of a *pressure cooker* model completes the presentation.



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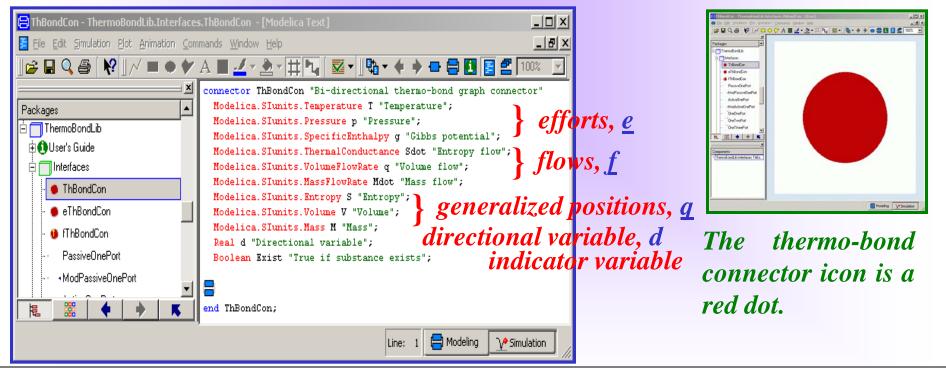
- <u>Thermo-bond graph connectors</u>
- <u>A-causal and causal bonds</u>
- <u>Bus-junctions</u>
- <u>Heat exchanger</u>
- Volume work
- Forced volume flow
- <u>Resistive field</u>
- <u>Pressure cooker</u>
- <u>Capacitive fields</u>

- Evaporation and condensation
- <u>Simulation of pressure cooker</u>
- Free convective mass flow
- Free convective volume flow
- Forced convective volume flow
- <u>Water serpentine</u>
- <u>Biosphere 2</u>



The Thermo-Bond Graph Connectors I

• We shall need to introduce new *thermo-bond graph connectors* to carry the six variables associated with the three flows. They are designed as an *11-tuple*.





 $\langle \Box \Box \rangle$

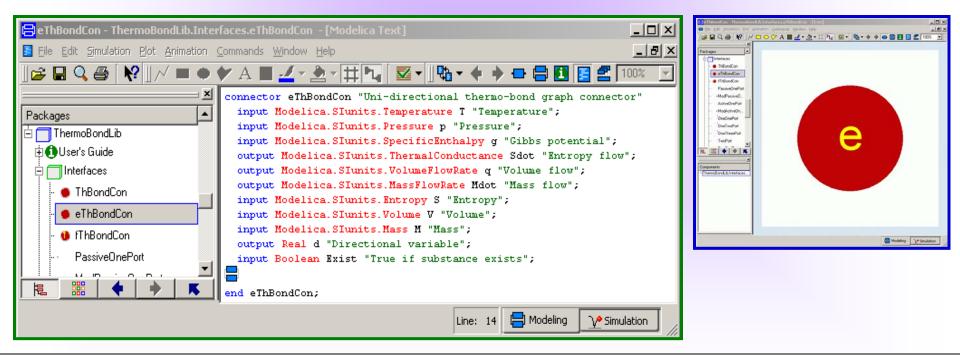
The A-Causal Thermo-Bond Model

The thermo-bond is equivalent to the regular bond, except that ten variables need to be connected across, instead of only two.	<pre> ThermoBond - ThermoBondLib.Bonds.ThermoBond - [Modelica Text] File Edit Simulation Plot Animation Commands Window Help Packages ThermoBondLib equation ThermoBondLib equation ThBondCon2.T = ThBondCon1.T; ThBondCon2.p = ThBondCon1.g; ThBondCon2.g = ThBondCon1.g; ThBondCon2.g = ThBondCon1.g; ThBondCon2.s = ThBondCon1.s; ThBondCon2.s = ThBondCon1.s; ThBondCon2.s = ThBondCon1.s; ThBondCon2.s = ThBondCon1.s; ThBondCon2.t = ThBondCon1.s; ThBondCon2.t = ThBondCon1.s; ThBondCon2.t = ThBondCon1.t; ThBondCon2.t = ThBondCon1.t = ThBondCon1.t = ThBondCon1.t = ThBondCon1.t = ThBondCon2.t = ThBondCon1.t = ThBondCon1.t = ThBondCon1.t = ThBondCon1.t = ThBondCon1.t = ThBondCon2.t = ThBondCon2.t = ThBondCon1.t = ThBondCon2.t = ThBond</pre>
November 22, 2012	Start Presentation

November 22, 2012

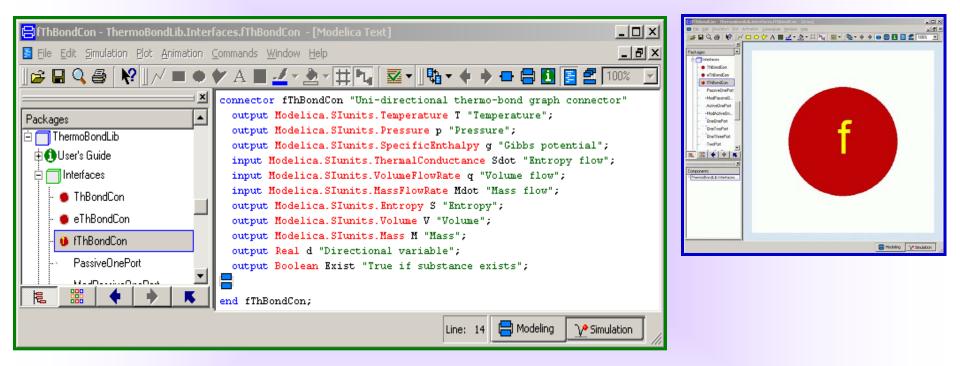
The Thermo-Bond Graph Connectors II

• Like in the case of the general bond-graph library, also the thermo-bond-graph library offers *causal* next to *a-causal* bonds.





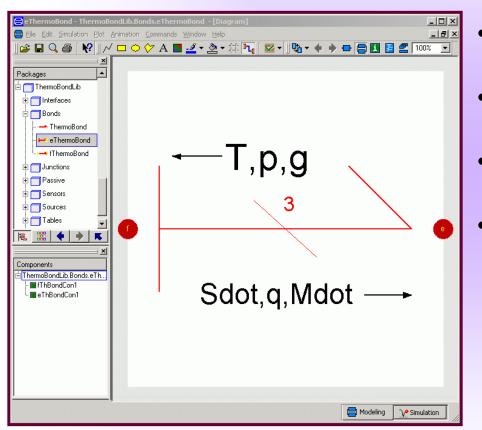
The Thermo-Bond Graph Connectors III



• Either the three efforts or the three flows are treated as input variables.



The Causal Thermo-Bond Blocks



- Using these connectors, causal thermo-bond blocks can be defined.
- The *f-connector* is used at the side of the causality stroke.
- The *e-connector* is used at the other side.
- The causal thermo-bond-graph connectors are only used in the context of the thermo-bond blocks.
 Everywhere else, the a-causal thermo-bond-graph connectors are to be used.



The Bus-0-Junctions

😑 ThreePortZero - ThermoBondLib.Interfaces.ThreePortZero - [Modelica Text] The junctions can now be programmed. Let us 🔄 File Edit Simulation Plot Animation Commands Window Help look at a bus-0-junction with three bond 'A 🔳 🚣 🔹 🖄 🕇 💾 🚺 🚰 🔲 🔍 🚑 📢 🛛 📈 🔳 🌒 × model ThreePortZero attachments Packages ThermoBondLib 🗄 🚯 User's Guide # B Q @ W // C O V A . 2 . 2 . 4 h 2 . 🗄 🥅 Interfaces 🐞 ThBondCon Lington lature. CosTHOPO eThBondCon ConThenelise TwoPost fThBondCon 1 Xad 1 Xad 1 Xad 1 Xad 1 Xad 1 Xad TeoPot2eo NodTwoPort **PassiveOnePort** TweePorCiro FourPartZero EvePatZen ModPassiveOnePort S + + 5 equation report for ActiveOnePort T[1] = ThBondConl.T; hemoBondLb1ne 1hBondCon1 1hBondCon2 1hBondCon2 p[1] = ThBondConl.p; ModActiveOnePort g[1] = ThBondConl.g; OneOnePort Modeling Ve Serulation B Holder Veterate OneTwoPort OneThreePort - 🗆 × 😑 J0p3 - ThermoBondLib.Junctions.J0p3 - [Modelica Text] S[1] = ThBondConl.S; V[1] = ThBondConl.V; TwoPort 🧮 File Edit Simulation Plot Animation Commands Window Help _ 8 × M[1] = ThBondConl.M; TwoPortZero <u>...</u> + + + A × • 🚅 🔲 🔍 T[2] = ThBondCon2.T; ModTwoPort **₽**Ъ -100% p[2] = ThBondCon2.p; ThreePortZero g[2] = ThBondCon2.g; × model JOp2 FourPortZero exten s Interfaces.ThreePortZero; . Packages FourPortZero2 🗄 🥅 Junctions FivePortZero S[2] = ThBondCon2.S; eguation V[2] = ThBondCon2.V; ...0 J0p11 T[2:3] = T[1:2];SixPortZero M[2] = ThBondCon2.M; p[2:3] = p[1:2];- <mark>0</mark> J0⊳12 EightPortZero g[2:3] = g[1:2];T[3] = ThBondCon3.T; - 0 J0p13 🗄 🥅 Bonds sum(Sdot) = 0;p[3] = ThBondCon3.p; sum(q) = 0;Junctions --<mark>0</mark> -J0_D2 g[3] = ThBondCon3.g; sum(Mdot) = 0;🗄 🥅 Passive 0 J0p3 S[2:3] = S[1:2];

רא Simulation

V[2:3] = V[1:2];

M[2:3] = M[1:2];

Line: 1

end JOp3;

Exist[2:3] = Exist[1:2];

Hodeling

--0 JOp4

- 0 - JOo5

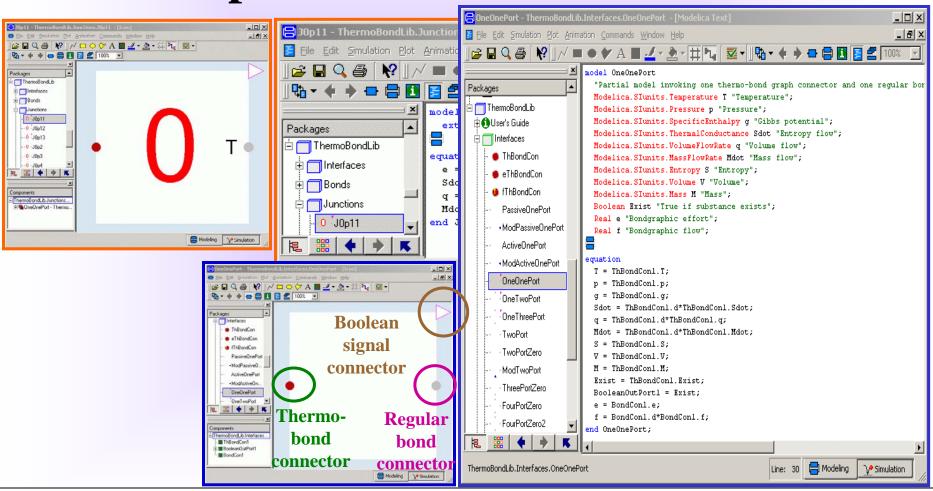
挹

🔽 🗸 🛯 🗞 🗸 🌲 📥 🖨 🚺 📑 🕰 100% "Partial model invoking three thermo-bond graph connectors" Modelica.SIunits.Temperature T[3] "Temperature"; Modelica.SIunits.Pressure p[3] "Pressure"; Modelica.SIunits.SpecificEnthalpy g[3] "Gibbs potential"; Modelica.SIunits.ThermalConductance Sdot[3] "Entropy flow"; Modelica.SIunits.VolumeFlowRate g[3] "Volume flow"; Modelica.SIunits.MassFlowRate Mdot[3] "Mass flow"; Modelica.SIunits.Entropy S[3] "Entropy"; Modelica.SIunits.Volume V[3] "Volume"; Modelica.SIunits.Mass M[3] "Mass"; Boolean Exist[3] "True if substance exists"; Sdot[1] = ThBondConl.d*ThBondConl.Sdot; q[1] = ThBondConl.d*ThBondConl.q; Mdot[1] = ThBondConl.d*ThBondConl.Mdot; Exist[1] = ThBondConl.Exist; Sdot[2] = ThBondCon2.d*ThBondCon2.Sdot; g[2] = ThBondCon2.d*ThBondCon2.g; Mdot[2] = ThBondCon2.d*ThBondCon2.Mdot; Exist[2] = ThBondCon2.Exist; Sdot[3] = ThBondCon3.d*ThBondCon3.Sdot; q[3] = ThBondCon3.d*ThBondCon3.q; Sensors 🦳 Mdot[3] = ThBondCon3.d*ThBondCon3.Mdot; S[3] = ThBondCon3.S; Sources V[3] = ThBondCon3.V;M[3] = ThBondCon3.M; Exist[3] = ThBondCon3.Exist; 📇 Modeling Line: 3 **Λ**^ρ Simulation

- 🗆 ×

_ 8 ×

Special Bus-0-Junctions I

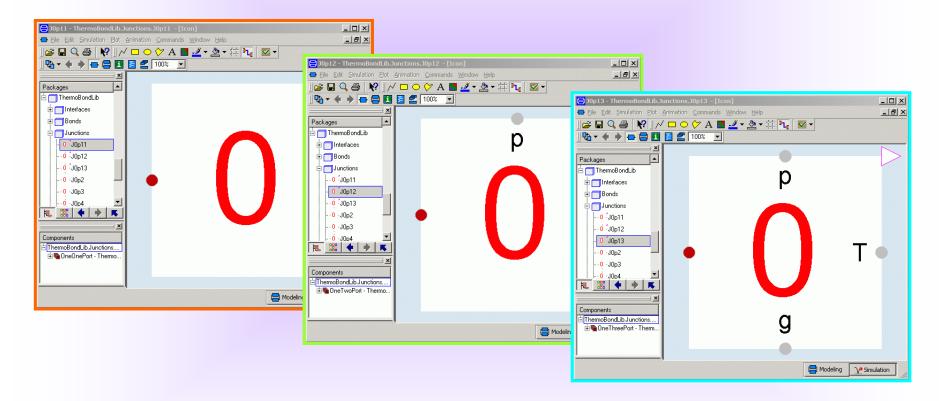


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Special Bus-0-Junctions II

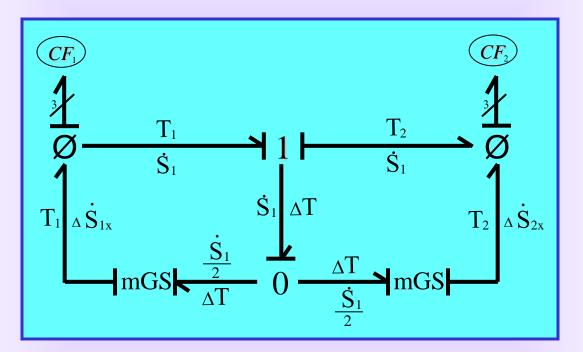


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The Heat Exchanger



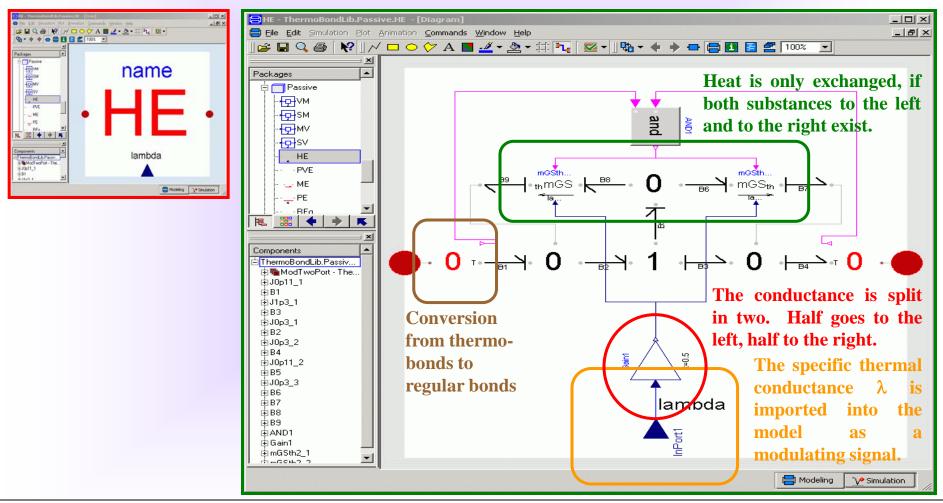
$$CF_{i} \xrightarrow{3} HE \xrightarrow{3} CF_{i}$$

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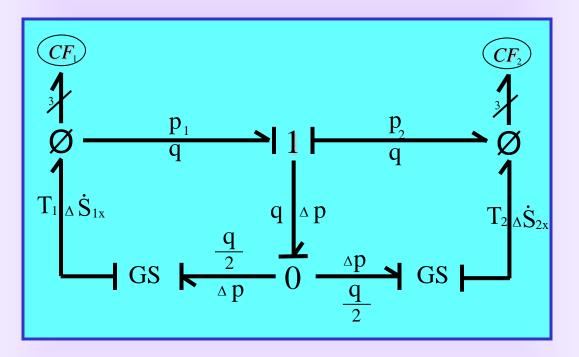
The Thermo-Bond Heat Exchange Element



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The Volume Work



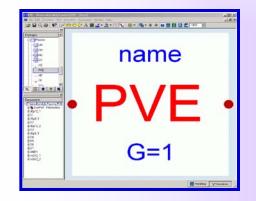
$$CF \xrightarrow{3} PVE \xrightarrow{3} CF$$

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The Pressure Volume Exchange Element

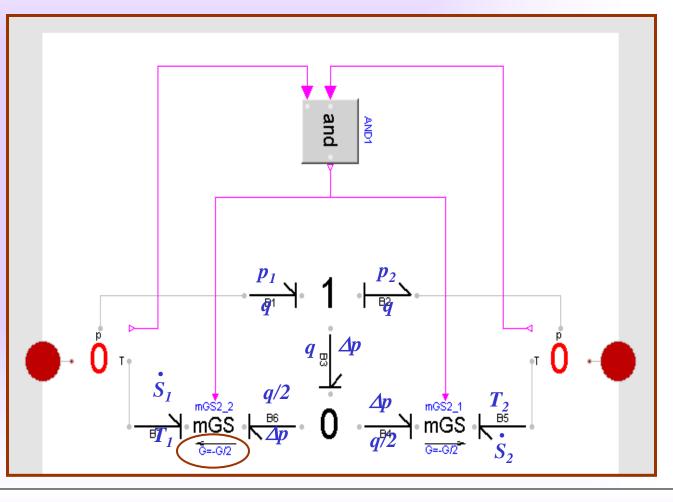


We pick up the two pressures.

The difference between the two pressures causes a negative volume flow.

The surplus power is split in two to be converted into additional entropy.

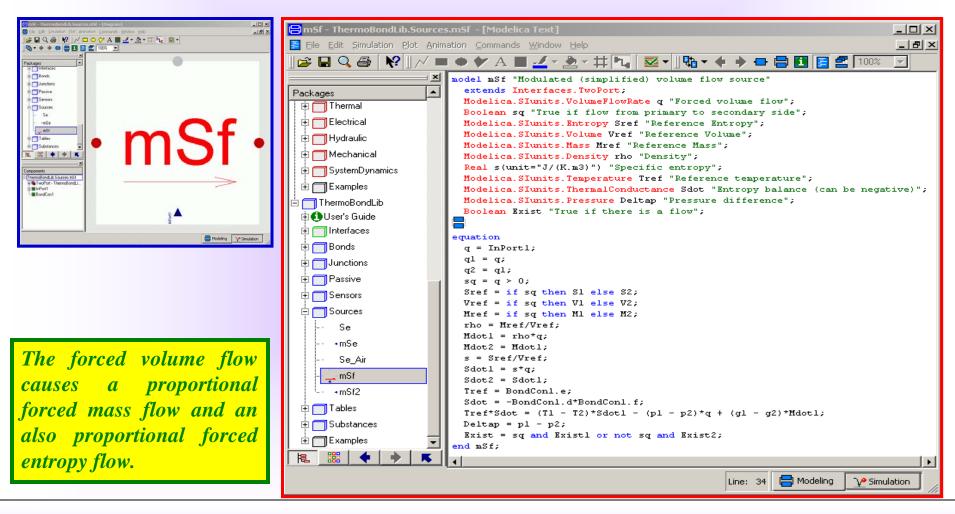
The generated entropy is delivered back to the thermal ports.





Start Presentation

Forced Volume Flow I



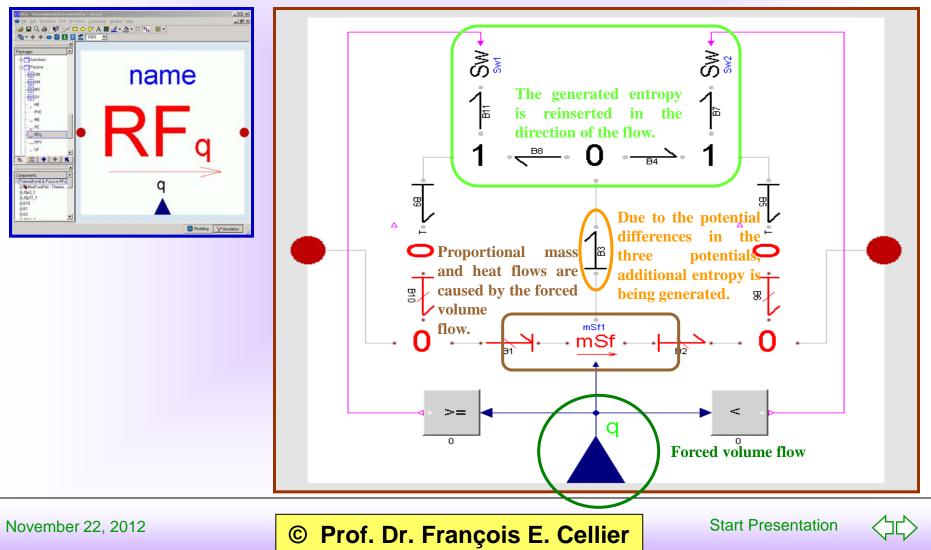
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Forced Volume Flow II

- The model presented here cannot yet be used to represent e.g. a *pump* or a *compressor*, because it doesn't consider the power needed to move the fluid around.
- The model is acceptable to describe small mass movements such as pressure equilibrations between the bulk and a (mathematical) boundary layer.
- An improved forced volume flow model shall be discussed later in this lecture.

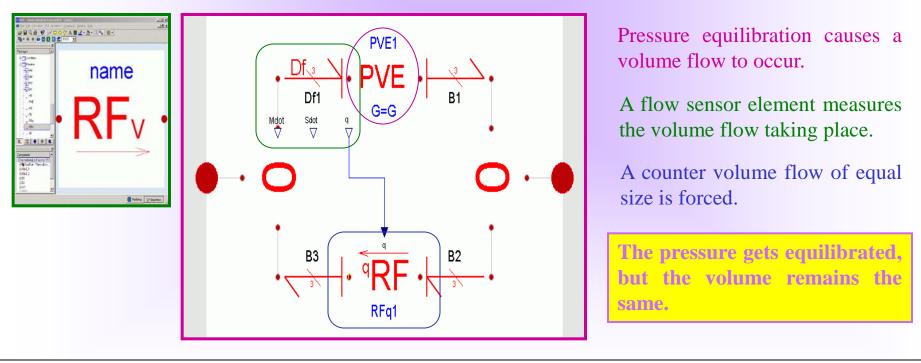


The Resistive Field



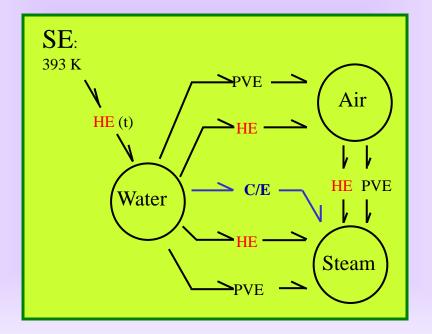
Pressure Equilibration With Constant Volume

• Sometimes it is useful to allow a mass flow to take place, while the volume doesn't change (remember the gas cartridge).





The Pressure Cooker I

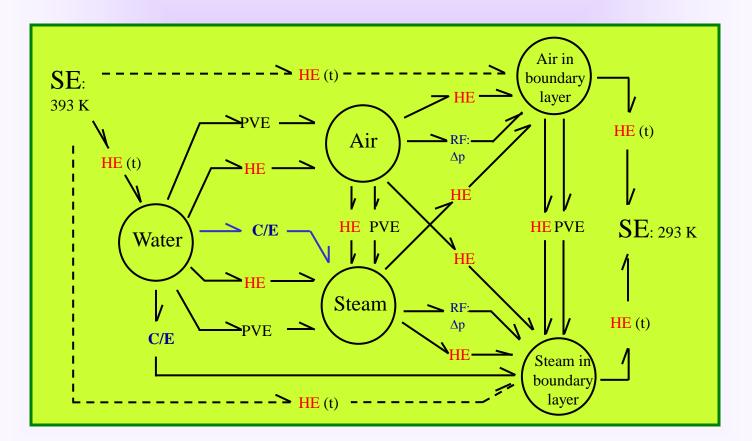


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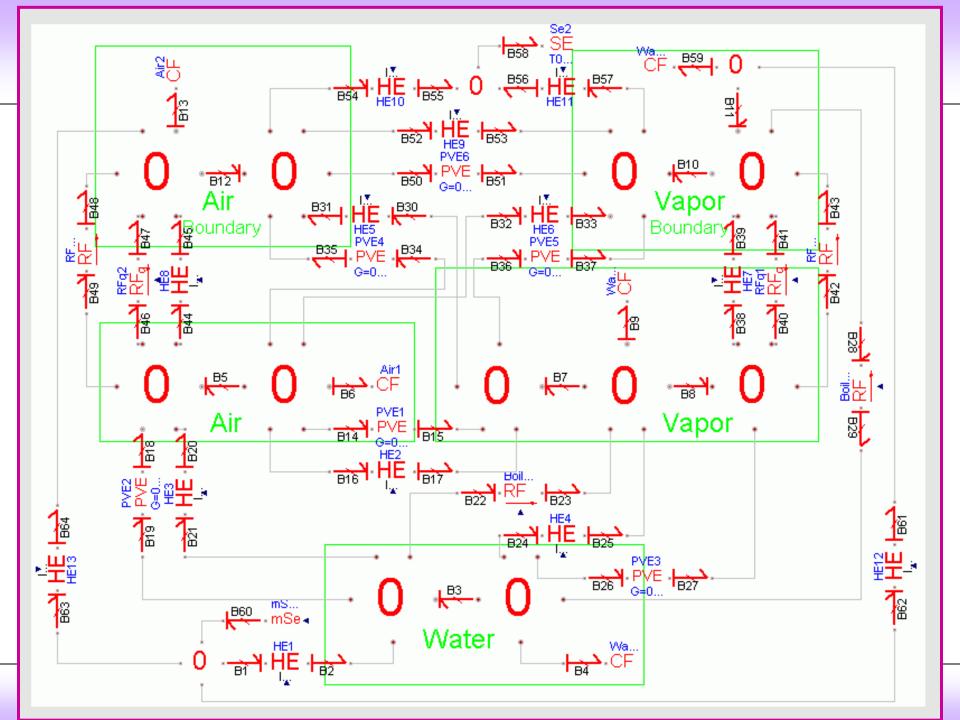
The Pressure Cooker II



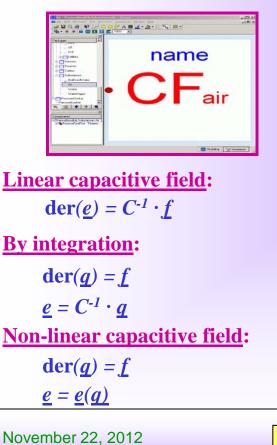
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• Let us look at the capacitive field for air.

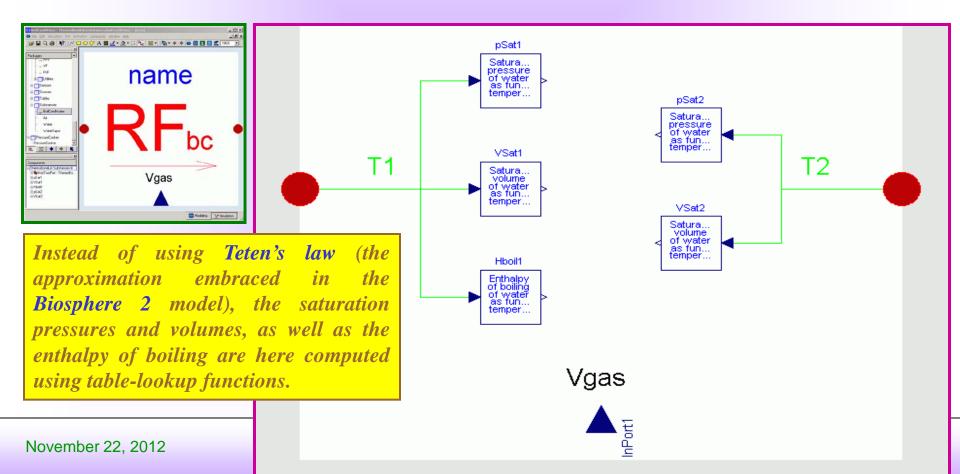


Capacitive Fields

😑 Air - ThermoBondLib.Substan	ces.Air - [Modelica Text]
File Edit Simulation Plot Anim	
	model Air "Capacitive field representing air"
Packages A	extends Interfaces.PassiveOnePort;
	parameter Modelica.Slunits.Entropy S0=6.81010184 "Entropy if no air";
	<pre>parameter Modelica.SIunits.Volume V0=0.83112221e-3 "Volume if no air";</pre>
🕀 🗖 Spice	parameter Modelica.SIunits.Mass MO=le-3 "Mass if no air";
🗄 🥅 Switches	parameter Modelica.SIunits.SpecificHeatCapacity cp=1004.0
■ Thermal	"Heat capacity of air at constant pressure"; parameter Modelica.SIunits.SpecificHeatCapacity R=287.2 "Gas constant";
	parameter Modelica.Slunits.Mass epsM=0.5e-6
	"Smallest mass distinguishable from zero";
🗄 🔲 Hydraulic	parameter Boolean fict=false "True is fictitious values are used";
🗄 🗇 Mechanical	parameter Modelica.SIunits.Temperature T_fict=298.53
⊕ ☐ SystemDynamics	"Fictitious temperature is no air";
	parameter Modelica.SIunits.Pressure p_fict=le5 "Fictitious pressure if no air";
	Modelica.SIunits.Entropy S int "Entropy of air";
🗄 🔲 ThermoBondLib	Modelica.SIunits.Volume V int "Volume of air";
🗄 🕕 User's Guide	Modelica.SIunits.Mass M_int "Mass of air";
🕂 🗖 Interfaces	Modelica.SIunits.SpecificHeatCapacity cv
Bonds	"Heat capacity of air at constant volume";
	Modelica.SIunits.SpecificVolume v "Specific volume"; Modelica.SIunits.SpecificEntropy s "Specific entropy";
	Real ln v "Natural logarithm of specific volume";
🗄 🗖 Sensors	equation
	$der(M_{int}) = Mdot;$
Tables	$\frac{\operatorname{der}(\underline{n}_{\operatorname{int}}) = \operatorname{Maot}}{\operatorname{der}(\underline{s}_{\operatorname{int}}) = \operatorname{Sdot}} \operatorname{der}(\underline{q}) = \underline{f}$
	der(V_int) = q;
E Substances	Exist = $M_{int} > epsM;$ cv = cp - R;
BoilCondWater	v = if Exist then V_int/M_int else 0;
- Air	<pre>s = if Exist then S_int/M_int else 0;</pre>
Air_isochoric	<pre>ln_v = Modelica.Math.log(V_int/M_int);</pre>
Water	$p = if Exist or not fict then T*R*M_int/V_int else p_fict;$ T = if Exist or not fict then 202 [Eterm(/z = 6012 Z = Bt/) = x + 0.17246))
e = e(a)	T = if Exist or not fict then 293.15*exp((s - 6813.7 - R*(ln_v + 0.17245)) /cv) else T fict;
- Water_isochetie	$g = T^{*}(cp - s);$
🦾 WaterVapor	M = if Exist then M_int else MO;
Examples 🚽	V = if Exist then V_int else V0;
	S = if Exist then S_int else SO;
	end Air;
0	Line: 41 📑 Modeling 📝 Simulation
<u> </u>	

Evaporation and Condensation I

• The models describing *evaporation* and *condensation* are constructed by interpolation from steam tables.



Evaporation and Condensation II

model BoilCondWater "Boiling and condensation of water"						
extends Interfaces.ModTwoPort(InPort equation						
Modelica.SIunits.Volume));	pSatl.u = Tl;					
<pre>parameter Real Rb(unit="m.s") = 0.0.</pre>	$p_{sat2.u} = T_2;$					
<pre>parameter Real Rc(unit="m.s") = 0.0.</pre>	VSatl.u = Tl;					
Modelica.SIunits.Pressure pSat_liq '	VSat2.u = T2;					
Modelica.SIunits.Pressure pSat_gas '	Hboill.u = Tl;					
Modelica.SIunits.SpecificVolume VSat						
Modelica.SIunits.SpecificVolume VSat	at pSat_gas = pSat2.y;					
Modelica.SIunits.SpecificEnthalpy h	h VSat_liq = VSatl.y;					
Modelica.SIunits.Volume Vgas "Gas vo	Iunits.Pressure pgas "Part hboil = Hboill.y;					
Modelica.SIunits.Pressure pgas "Part						
Modelica.SIunits.MassFlowRate Mboil	SIunits.MassFlowRate Mboil					
Modelica.SIunits.MassFlowRate Mcond	nd pgas = p2*V2/Vgas;					
Boolean fd "True if forward flow";	Mboil = if ((pSat_liq > pl) and Exist) then Rb*(pSat_liq - pl) else 0;					
Modelica.SIunits.HeatFlowRate Delta	Mcond = if ((pSat_gas < pgas) and Exist2) then Rc*(pgas - pSat_gas) else al					
"Heat generated by potential equil	equi Mdotl = Mboil - Mcond;					
Boolean Exist "True if flow exists";	s"; Mdot2 = Mdot1; eno ql = Mboil*VSat_liq - Mcond*VSat_gas;					
Modelica.SIunits.Mass Mref "Referend						
Modelica.SIunits.AngularFrequency x:						
Modelica, SIunits, ThermalConductance	1- 1-7					
Modelica.SIunits.ThermalConductance	DeltaQdot = (gl - g2)*Mdotl - (pl - p2)*gl;					
Modelica.SIunits.Volume Vsteam "Max:	Exist = (fd and Existl) or (not fd and Exist2);					
Real huml "Relative humidity compute	Mref = if fd then M1 else M2;					
Real hum2 "Relative humidity compute	<pre>xi = if Exist then Mdotl/Mref else 0;</pre>					
	Sdot2_aux = if Exist2 then Sdot1 + (DeltaQdot + (T1 - T2)*Sdot1)/T1 else					
	Sdotl;					
	Sdotl_aux = if Existl then Sdot2 + (DeltaQdot + (T1 - T2)*Sdot2)/T2 else					
A bit messy!	Sdot2;					
5	<pre>Sdotl = if fd then xi*Sl + hboil*Mdotl/Tl else Sdotl_aux;</pre>					
	Sdot2 = if fd then Sdot2_aux else xi*S2 - hboil*Mdot2/T2;					
	Vsteam = 1/(p2/pSat_gas - 1)*Vgas;					
November 22, 2012	huml = pgas/pSat_gas; hum2 = 1 - V2/Vsteam;					
	num2 = 1 - V2/Vsteam; end BoilCondWater;					
	ena bolicondwater,					

Simulation of Pressure Cooker

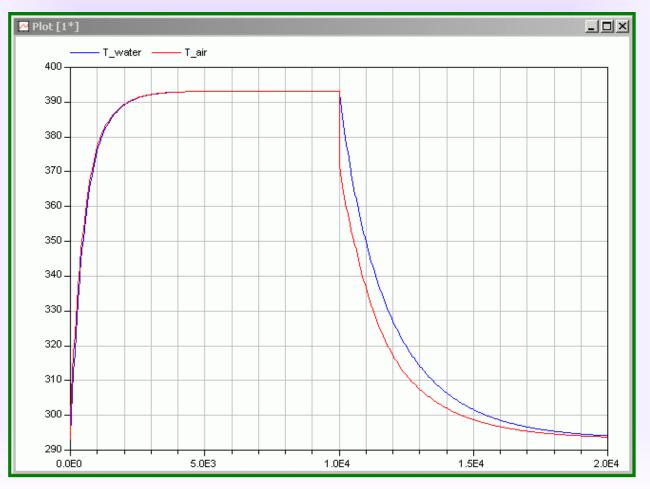
• We are now ready to compile and simulate the model.

🖨 Messages - Dymola 📃		Experiment Setup	🖨 Messages - Dymola 📃 🔳 🗙
Syntax Error Translation Dialog Error Simulation Translation of ThermoBondLib Examples PressureLooker: DAE having 10556 scalar unknowns and 10556 scalar equations. STATISTICS Original Model Number of components: 625 Variables: 9605 Constants: 0 Parameters: 179 (733 scalars) Unknowns: 9426 (10556 scalars) Differentiated variables: 15 scalars Equations: 6387 Nontrivial: 3745 Translated Model Constants: 2248 scalars Prepending: 31 scalars Input: 0 Outputs: 14 scalars Parameter: 774 scalars Parameter: 78 scalars Inputs: 0 Outputs: 14 scalars Continuous time states: 15 scalars Ime-vaying variables: 7851 calars Alias variables: 7851 calars Assumed default initial conditions=true; gives more information Number of mixed real/discrete systems of equations: 0 Sizes of linear systems of equations: (4, 4) Sizes of inear systems of equations: (6, 6, 6, 6) Size	٩ •	General Iranslation Qutput Debug Compiler Realtime Experiment Name PressureCooker Image: Compiler Realtime Image: Compiler Realtime Simulation interval Simulation interval Image: Compiler Realtime Image: Compiler Realtime Start time Image: Compiler Simulation Image: Compiler Realtime Image: Compiler Realtime Start time Image: Compiler Simulation Image: Compiler Simulation Image: Compiler Image: Compile	Syntax ErrorTranslationDialog ErrorSimulationLog-file of program ./dymosim(generated: Thu Feb 07 23:33:49 2008)dymosin started "dsin.txt" loading (dymosim input file) "dsin.txt" loading (dymosim input file) "PressureCooker.mat" creating (simulation result file)Integration started at T = 0 using integration method DASSL(DAB multi-step solver (dassl/dasslrt of Petzold modified by Dynasim))Integration terminated successfully at T = 20000CPU-time for integration: 1.54 secondsCPU-time for one GRID interval: 5:00 milli-secondsMumber of result points: 522Mumber of GRID points: 15458Mumber of Jacobian-evaluations: 625Mumber of fully time events: 1Mumber of state events: 0Mumber of state events: 0Mumber of state: 1.9e-009Maximum integration stepsize: 1.9e-009Maximum integration order: 5Calling terminal section "dsfinal.txt" creating (final states)

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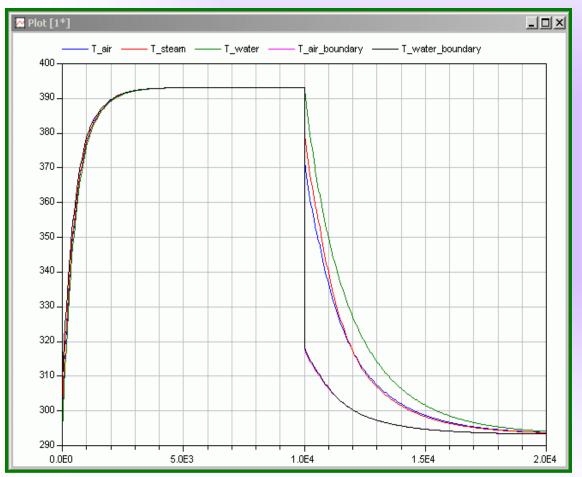
Simulation Results



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Simulation Results II



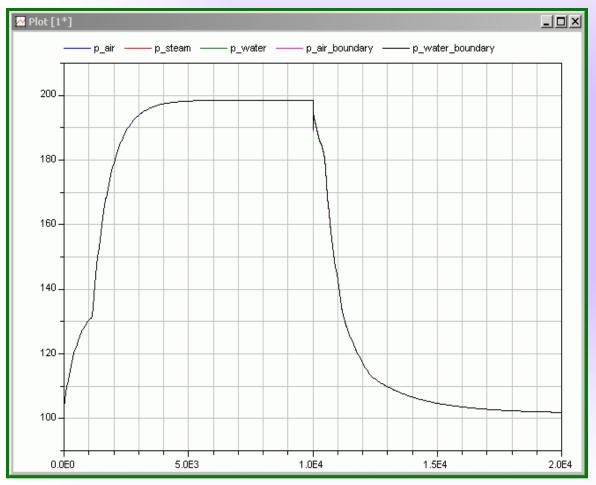
Heating is sufficiently slow that the temperature values of the different media are essentially indistinguishable. The heat exchangers have a smaller time constant than the heating.

During the cooling phase, the picture is very different. When cold water is poured over the pressure cooker, air and steam in the small boundary layer cool down almost instantly. Air and steam in the bulk cool down more slowly, and the liquid water cools down last.

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Simulation Results III



The pressure values are essentially indistinguishable throughout the simulation.

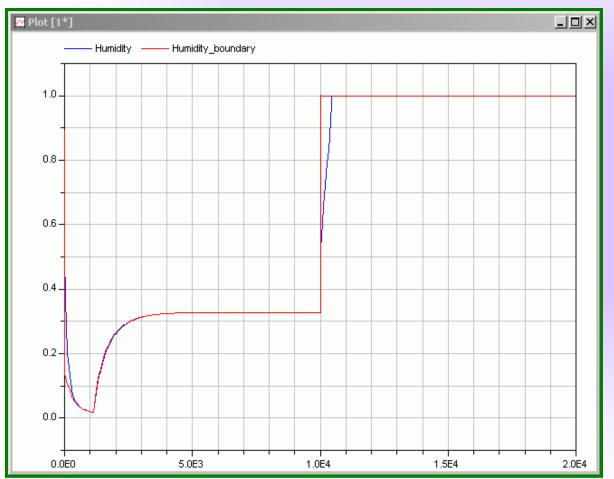
During the heating phase, the pressures rise first due to rising temperature. After about 150 seconds, the liquid water begins to boil, after which the pressure rises faster, because more steam is produced (water vapor occupies more space at the same temperature than liquid water).

The difference between boundary layer and bulk pressure values in the cooling phase is a numerical artifact.

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Simulation Results IV



The relative humidity decreases at first, because the saturation pressure rises with temperature, i.e., more humidity can be stored at higher temperatures.

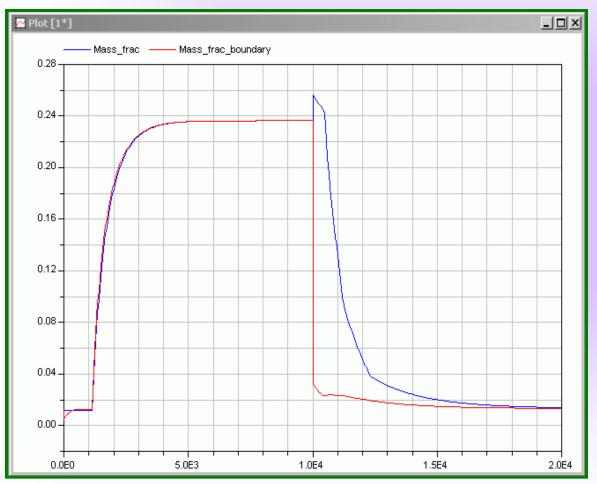
As boiling begins, the humidity rises sharply, since additional vapor is produced.

In the cooling phase, the humidity quickly goes into saturation, and stays there, because the only way to ever get out of saturation again would be by reheating the water.

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Simulation Results V



The *mass fraction* defines the percentage of water vapor contained in the air/steam mixture.

Until the water begins to boil, the mass fraction is constant. It then rises rapidly until it reaches a new equilibrium, where evaporation and condensation balance out.

During the cooling phase, the boundary layer cools down quickly, and can no longer hold the water vapor contained. Some falls out as water, whereas other steam gets pushed into the bulk, temporarily increasing the mass fraction there even further.

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Free Convective Mass Flow

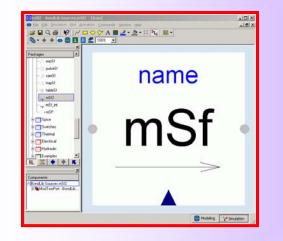
- We are now ready to discuss *free convective mass flow*, such as mass flow occurring in a segment of a pipe.
- The convective mass flow occurs because more mass is pushed in from one end, pushing the mass currently inside the pipe segment out by the other end.
- To this end, we need to introduce some more models.

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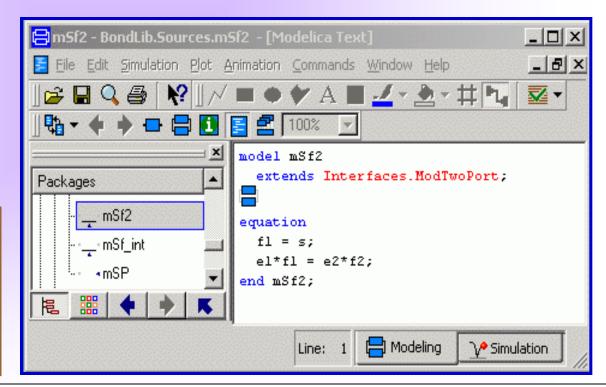


The Forced Flow Source

• This model describes an element of the regular bond-graph library.

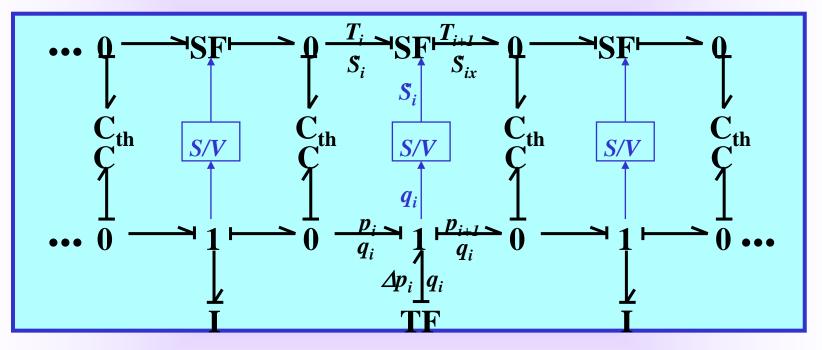


The primary side is a flow source, the secondary side can be either a flow or an effort source. Its equation is defined to satisfy power continuity across the element.





Density and Specific Entropy I

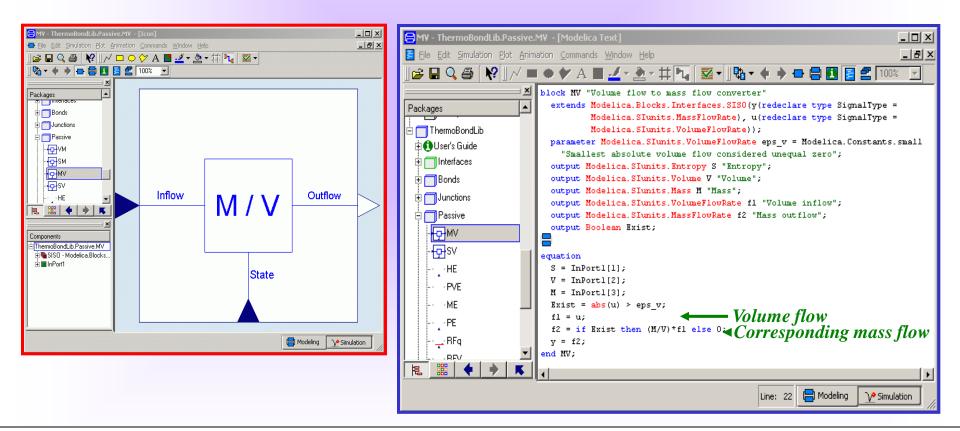


• As mentioned some lectures ago, we shall need modulated flow sources (as introduced one slide ago) that are modulated by the specific entropy and/or the specific mass (i.e., the density).



Density and Specific Entropy II

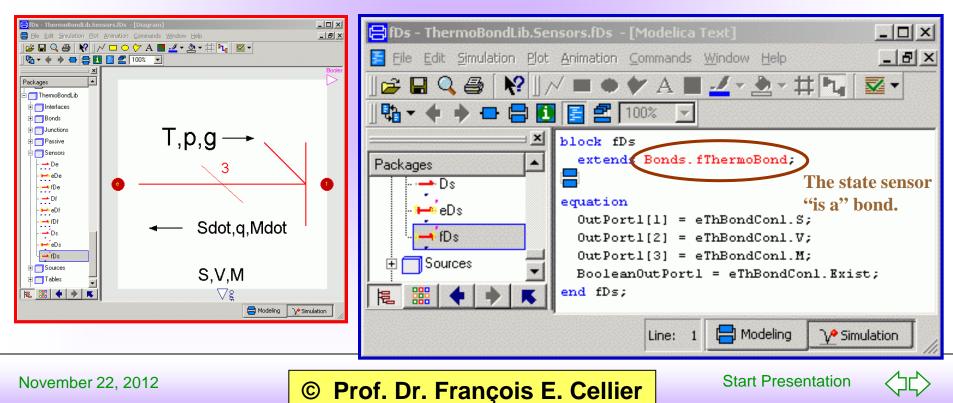
• These models are created as blocks:





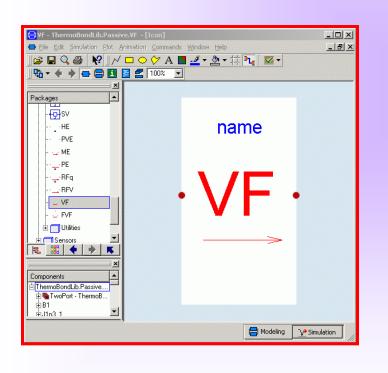
The State Sensor

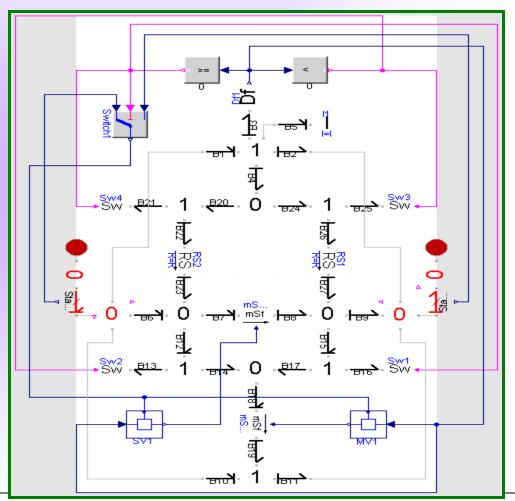
• Many elements that are related to substances require state information. This is generated by a specialized thermo-bond, the so-called *state sensor element*.



Free Convective Volume Flow

• We are now ready to describe the free convective volume flow.

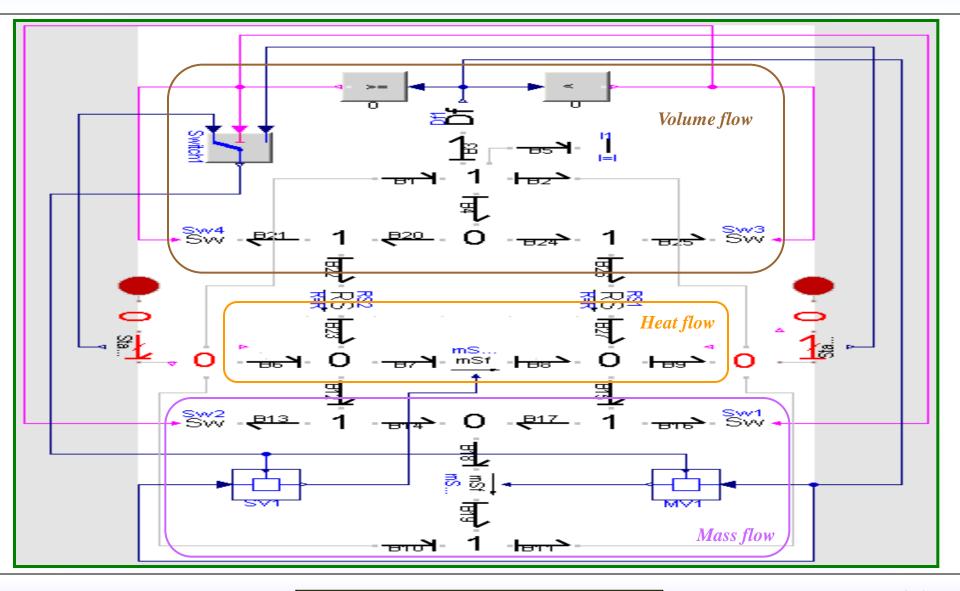








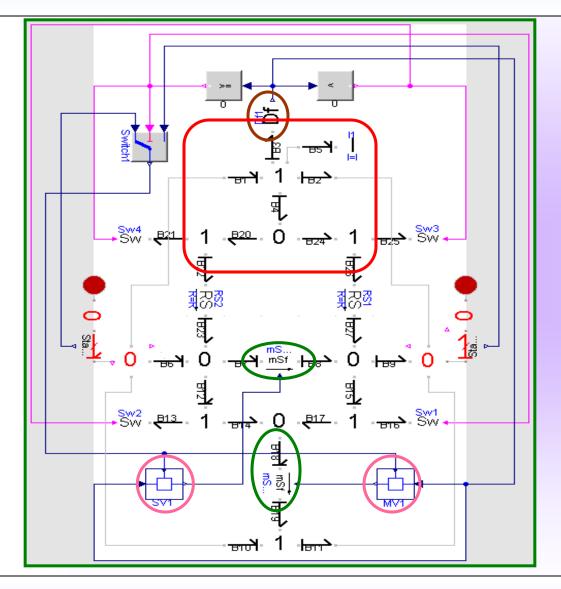
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ETH

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The volume flow is modeled as a wave equation with friction. The friction is in parallel with the inertia.

The flow is measured using a flow sensor element. The additional entropy generated by friction is reintroduced in the down-wind direction, i.e., in the direction of the flow. Switch elements are used to determine the reintroduction point.

Non-linear flow sources are used to model the parallel thermal and mass flows.

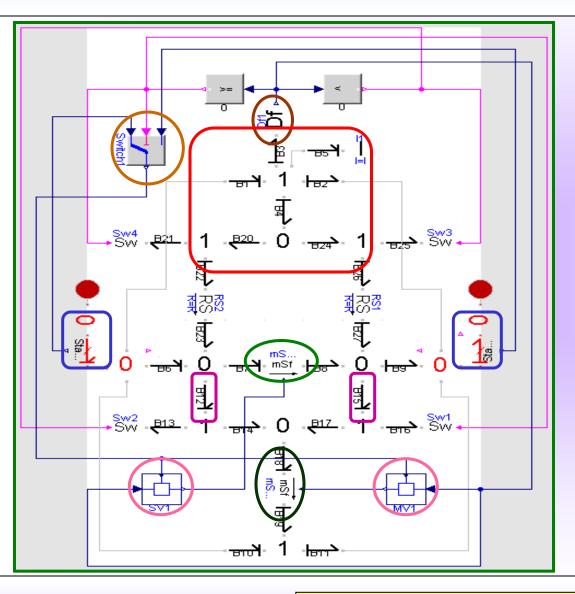
These are computed by converting the volume flows to consistent entropy and mass flows.

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ETH

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State sensor elements are used to determine the current values of volume, entropy, and mass.

Up-wind state information is being used to convert the volume flow into consistent entropy and mass flows.

Since entropy doesn't need to be preserved, the non-linear flow source is inserted directly into the thermal branch.

Since mass flow must be preserved, the non-linear flow source is inserted under a 1-junction in the mass flow branch.

To move mass with the volume, additional energy is needed that is taken from the thermal domain.

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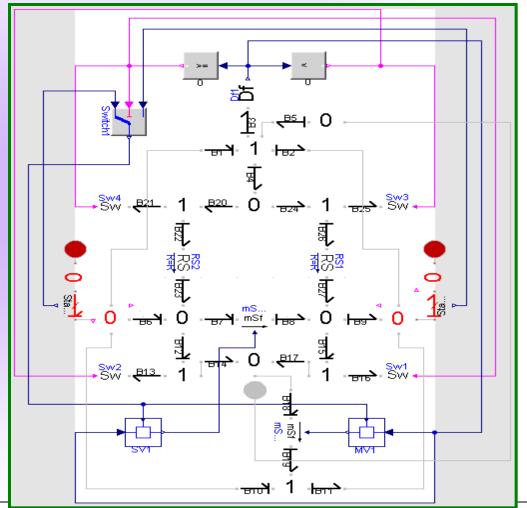


Forced Convective Volume Flow

• We are now ready to describe the forced convective mass flow.

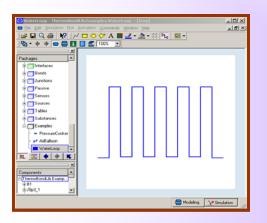


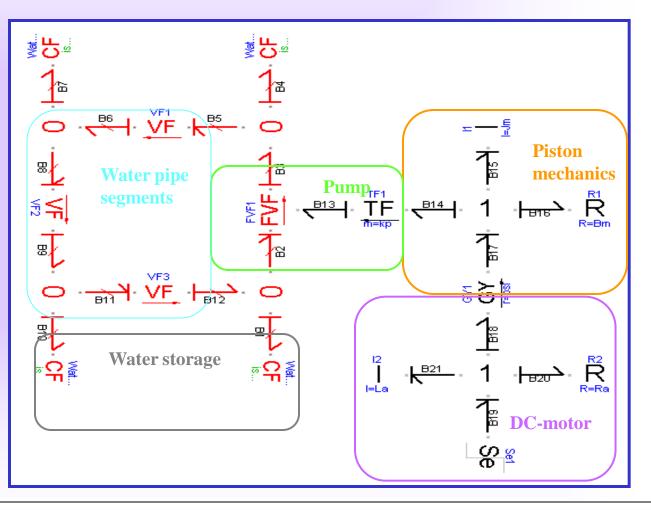
The model is almost the same as the free convective flow model, except that a volume flow is forced on the system through the regular bond connector at the top, replacing the inductor.





The Water Serpentine I





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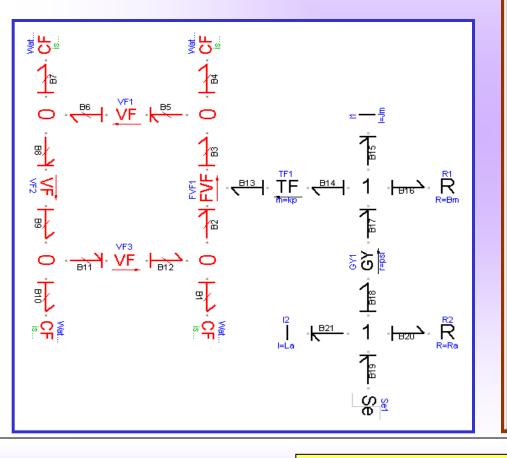
The Water Serpentine II

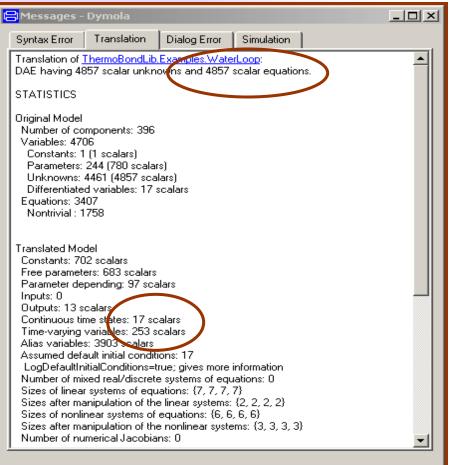
- The pump forces a flow, thereby creating a higher pressure at the outflow, while creating a lower pressure at the inflow.
- Mass is transported through the pump with the volume. Since the mass is getting condensed, it occupies less space. Thus, there is "surplus" volume that gets used to "finance" the mass transport in the *Gibbs equation*.
- In the pipe segments, the pressure is gradually reduced again, thus each pipe segment has a higher pressure at the inflow than at the outflow. The mass thus expands, and the volume consumed in the pump is gradually given back, so that the overall volume in the water serpentine is being preserved.



The Water Serpentine III

• We are now ready to simulate this system.







The Water Serpentine IV

Amessages - Dymola	
Syntax Error Translation Dialog Error Simulation	p1p2p3p4
Log-file of program ./dymosim (generated: Fri Feb 08 16:13:54 2008) dymosim started "dsin.txt" loading (dymosim input file) "WaterLoop.mat" creating (simulation result file) Integration started at T = 0 using integration method DASSL (DAE multi-step solver (dassl/dasslrt of Petzold modified by Dynasim); Integration terminated successfully at 1 = 4 CPU-time for integration : 0.4 seconds CPU-time for one GRID interval: 0.8 milli-seconds Number of result points : 514	
Number of GRID points : 501 Number of (successful) steps : 499	
Number of F-evaluations : 3336 Number of H-evaluations : 1053 Number of Jacobian-evaluations: 138 Number of (model) time events : 1 Number of (U) time events : 0 Number of state events : 6 Number of step events : 0 Minimum integration stepsize : 6.17e-009 Maximum integration stepsize : 0.476 Maximum integration order : 5 Calling terminal section "dsfinal.txt" creating (final states)	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

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Comparison With Biosphere 2

- In the *Biosphere 2* model, only the (sensible and latent) heat were modeled. The mass flows were not considered.
- Consequently, you never know in the *Biosphere 2* model, how much water is available where. It is always assumed that the pond never dries out, and that the plants always have enough water to be able to evaporate in accordance with their temperature and saturation pressure.
- In the case of the *pressure cooker* model, both the mass flows and the heat flows were modeled and simulated. Consequently, the case is caught, where all the water has evaporated, while the air/steam mixture is still not fully saturated.



References I

- Cellier, F.E. and J. Greifeneder (2003), "Object-oriented modeling of convective flows using the Dymola thermobond-graph library," *Proc. ICBGM'03, Intl. Conference on Bond Graph Modeling and Simulation*, Orlando, FL, pp. 198 – 204.
- Greifeneder, J. and F.E. Cellier (2001), "<u>Modeling multi-element systems using bond graphs</u>," *Proc. ESS'01, European Simulation Symposium*, Marseille, France, pp. 758 766.
- Cellier, F.E. and J. Greifeneder (2008), "<u>ThermoBondLib A New Modelica Library for Modeling Convective Flows</u>," *Proc. Modelica'08*, Bielefeld, Germany, pp. 163 – 172.



References II

- Cellier, F.E. (2007), *<u>The Dymola Bond-Graph Library</u>*, Version 2.3.
- Cellier, F.E. (2007), *<u>The Dymola Thermo-Bond-Graph</u> <u>Library</u>, Version 2.0.*

