

## ETH ETH Mathematical Modeling of Physical Systems Eidgenössische Technische Hochschule Zünich Swiss. Sedereil Institute ef Technology Zunich **Limitations of Bond Graphs I** • Bond graphs have been designed around the *conservation principles* of physics (energy conservation, mass conservation), and are therefore only suitable for the description of physical systems. • Chemistry was a border-line case. Although it is possible to model chemical reaction dynamics down to the level of physics, this is not truly necessary, since the reaction rate equations are decoupled from the energy balance equations. Hence this is rarely done. We did it, because the bond-graphic interpretation of chemical reactions offered additional insight that we could not have gained easily by other means. • Yet, as the complexity of molecules grows, especially in organic *chemistry*, it becomes more and more difficult to know what the elementary step reactions are, and at that level, chemistry becomes an *empirical science*, the knowledge of which is essentially covered by interpretations of observations alone.

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Ergenessische Technischenbornschule Zühlen Swiss-Redenal Inschulte af Technology Jurich	Mathematical Modeling of Physical Systems
<ul> <li>References</li> <li>Cellier, F.E. (1991), <u>Continuous System Modeling</u>, Springer-Verlag, New York, <u>Chapter 10</u>.</li> <li>Cellier, F.E. (2002), <u>Matlab code to curve-fit a logistic</u> model to the U.S. census data.</li> <li>Cellier, F.E. and A. Fischlin (1982), "<u>Computer-assisted</u> modeling of ill-defined systems," in: Progress in Cybernetics and Systems Research, Vol. 8, General Systems Methodology, Mathematical Systems Theory, Fuzzy Sets (R. Trappl, G.J. Klir, and F.R. Pichler, eds.), Hemisphere Publishing, pp. 417-429.</li> </ul>	
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